

A PROCESS FOR THE PREPARATION
OF OPTIONALLY SUBSTITUTED P-HYDROXYMANDELIC COMPOUNDS

A subject of the present invention is a process for the preparation of optionally substituted p-hydroxymandelic compounds and derivatives.

In the disclosure of the invention which follows, the term, "optionally substituted p-hydroxymandelic compounds" is used to refer to an aromatic compound which carries at least one $-CHOH-COOH$ group in para position of a hydroxyl group.

The present invention relates, more particularly, to the preparation of p-hydroxymandelic acid and 3-methoxy p-hydroxymandelic acid.

One of the conventional synthesis routes for p-hydroxymandelic acids consists in carrying out the condensation, in alkaline medium, of glyoxylic acid on phenol and/or its corresponding derivatives.

The yield is limited by the fact that the condensation reaction is not selective and also produces o-hydroxymandelic acids and dimandelic acids.

Furthermore, the reaction yield is reduced due to a parasitic secondary reaction. In fact, glyoxylic acid in aqueous alkaline medium is converted according to Cannizaro's reaction into oxalic and glycolic acids.

In order to prevent this Cannizaro's reaction from becoming dominant and destroying the glyoxylic acid, a proposal has been made in FR-A 2 132 364 for the condensation reaction to be carried out in a dilute aqueous medium and at low temperature or ambient temperature.

In view of the difficulty in obtaining satisfactory reaction yields, it is important to control the various parameters of the process, and, in particular, the quality of the glyoxylic acid used.

The most worthwhile process from an industrial viewpoint for the preparation of glyoxylic acid consists in oxidising the glyoxal with nitric acid. In this way, aqueous solutions of glyoxylic acid are obtained, which, in addition to unreacted glyoxal, also contain oxalic acid, organic acids, such as formic acid, acetic acid, glycolic acid, and nitric acid.

Until now, new methods have been constantly sought for the separation and purification of glyoxylic acid.

Therefore, a process was proposed in DE-A 1 198 339 which first and foremostly permitted the nitric acid to be eliminated, followed by the oxalic acid using basic ion exchange resins, followed by the glyoxal and the other impurities by over-concentration of the solution and crystallisation.

In DE-A 2 501 743, a process was disclosed in which the glyoxylic acid is separated from these impurities by extraction using aliphatic or cycloaliphatic alcohols, or aliphatic esters of alcohols with low carbon condensation.

A process for obtaining aqueous solutions of glyoxylic acid free from other acids has also been described in FR-A 2 552 426 which consists in treating the starting solution with an organic nitrogenous compound, preferably a tertiary amine, at a temperature which is at the most equal to 50°C, then in extracting the glyoxylic acid by extraction of the organic phase with water, at a higher temperature.

A constant concern has therefore been noted in the prior art of providing a solution of glyoxylic acid which is free of impurities.

Running counter to this teaching, it has been found that within the scope of preparing optionally substituted p-hydroxymandelic compounds condensation of glyoxylic acid and of the corresponding phenol is carried out with an increased yield, provided that said reaction is carried out in the presence of a dicarboxylic acid used in a certain quantity.

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A precise subject of the present invention is a process for the preparation of optionally substituted p-hydroxymandelic compounds and derivatives, which consists in carrying out the condensation, in water, in the presence of an alkaline agent, of an aromatic carrier compound

with at least one hydroxyl group and the para position of which is free, with glyoxylic acid, said process being characterised in that the reaction is carried out in the presence of an effective quantity of a carrier compound with at least two carboxylic functions.

In accordance with the process of the invention, the use of a catalyst according to the invention allows the reaction yield to be increased.

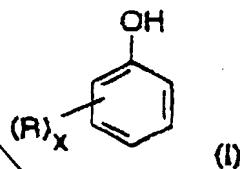
Another advantage of the process according to the invention is that it can involve a more technical glyoxylic acid containing, inter alia, oxalic acid.

The process according to the invention is used, most particularly, with phenol but also with substituted phenols which have at least one non-substituted para position.

The aromatic nucleus carries at least one hydroxyl group, but it can also carry one or more other substituents. Generally, "several substituents" means less than four substituents per aromatic nucleus.

Any substituent can be present provided that it is does not interfere in the reaction of the invention.

Therefore, the process according to the invention is well suited for use with hydroxylated aromatic compounds corresponding to the following formula (I):



in which formula (I):

- the para position is free,
- x is an integer between 1 and 4,
- R represents :
 - a hydrogen atom,

- a hydrocarbon group having from 1 to 20 carbon atoms selected from the alkyl, alkoxy, hydroxyalkyl, cycloalkyl, aryl, phenoxy, alkoxyalkyl, fluoroalkyl, hydroxyalkoxyalkylene groups,
- a hydroxyl group,
- a -CHO group,
- an acyl group having from 2 to 6 carbon atoms,
- a halogen atom, preferably a fluorine, chlorine or bromine atom,
- two R groups placed on two vicinal carbon atoms can form together and with the carbon atoms which carry them a benzene ring.

Examples of R radicals which are capable of being carried by the aromatic nucleus are given hereinafter:

- alkyl radicals, such as methyl, ethyl, propyl, isopropyl, butyl, isobutyl, sec-butyl, tert-butyl, n-octyl, 2-ethyl hexyl, decyl, octadecyl, eicosyl,
- alkoxy radicals, such as methoxy, ethoxy, propoxy, isopropoxy, butoxy, isobutoxy, sec-butoxy, tert-butoxy, hexyloxy, decyloxy, hexadecyloxy, octadecyloxy, or a phenoxy radical,
- hydroxyalkyl radicals, such as hydroxymethyl, hydroxyethyl, hydroxypropyl, hydroxyhexyl, hydroxydecyl,
- cycloalkyl radicals, such as cyclopentyl, cyclohexyl, cycloheptyl,
- fluoroalkyl radicals, such as fluoromethyl, difluoromethyl, trifluoromethyl, fluoroethyl, 1,1,1-trifluoro ethyl, pentafluoroethyl, fluoropropyl, fluorobutyl, trifluoroamyl,
- hydroxyalkoxyalkylene radicals, such as hydroxymethoxyethylene, hydroxyethyl di-(oxyethylene), hydroxyethyl tri-(oxyethylene), 1,2-hydroxyethoxypropylene, hydroxyethoxybutylene, hydroxypropoxypropylene, hydroxybutyloxybutylene, hydroxybutyl di-(oxybutylene),
- halogen atoms, such as fluorine, chlorine, bromine, or iodine.

Quite preferably, hydroxylated aromatic compounds are used in the process of the invention, corresponding to the general formula (I), in which:

- x is equal to 0, 1, 2 or 3,
- R represents one of the following groups or functions:
 - . a hydrogen atom,

- . a linear or branched alkyl radical having from 1 to 10 carbon atoms, and preferably from 1 to 4 carbon atoms,
- . a linear or branched alkoxy radical having from 1 to 10 carbon atoms, preferably from 1 to 4 carbon atoms,
- . an -OH group,
- . a -CHO group,
- . a halogen atom,
- . a -CF₃ group.

Still more preferably, the compounds of formula (I) are selected in which the R radicals which are identical or different are a hydrogen atom, a linear or branched alkyl radical with 1 to 4 carbon atoms, such as the methyl, ethyl, n-propyl, isopropyl, n-butyl or isobutyl radicals, a linear or branched alkoxy radical having 1 to 4 carbon atoms, such as the methoxy or ethoxy radicals, a -CHO group, or a chlorine atom, and x is preferably equal to 0 or 1.

By way of illustration of compounds corresponding to formula (I), the following can be mentioned:

- those corresponding to formula (I) in which x is equal to 0, such as phenol,
- those corresponding to formula (I) in which x is equal to 1, such as
 - . pyrocatechin
 - . resorcin
 - . o-cresol
 - . m-cresol
 - . 2-ethyl phenol
 - . 3-ethyl phenol
 - . 2-propyl phenol
 - . 2-sec-butyl phenol
 - . 2-tert-butyl phenol
 - . 3-tert-butyl phenol
 - . 2-methoxy phenol (guaiacol)
 - . 3-methoxy-phenol
 - . 2-ethoxy phenol (guetol)
 - . 2-isopropoxy phenol
 - . salicylic aldehyde

- . methyl salicylate
- . 2-chloro phenol
- . 3-chloro phenol
- . 3-nitro phenol
- those corresponding to formula (I) in which x is equal to 2, such as:
 - . 2,3-dimethyl phenol
 - . 2,5-dimethyl phenol
 - . 3,5-dimethyl phenol
 - . 2-hydroxy 5-acetamido benzaldehyde
 - . 2-hydroxy 5-ethamido benzaldehyde
 - . 2,3-dichloro phenol
 - . 2,5-dichloro phenol
 - . 3,5-dichloro phenol
 - . pyrogallol
- those corresponding to formula (I) in which x is equal to 3, such as:
 - . 2,3,5-trimethyl phenol
 - . 3,5-di-tert butyl phenol
 - . 2,3,5-trichloro phenol
- those corresponding to formula (I) having a naphthalenic radical, such as:
 - . 1-naphthol
 - . 2-naphthol
 - . 1,2-dihydroxy naphthalene
 - . 1,5-dihydroxy naphthalene
 - . 2,3-dihydroxy naphthalene
 - . 2,6-dihydroxy naphthalene
 - . 2,7-dihydroxy naphthalene
 - . 6-bromo 2-naphthol
- those corresponding to formula (I) having a chain formation of benzene nuclei:
 - . 2-phenoxy phenol
 - . 3-phenoxy phenol

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Of the list of afore-mentioned compounds, the aromatic carrier compounds preferably used which have at least one hydroxyl group are: phenol, o-cresol, m-cresol, 3-ethyl phenol, 2-tert-butyl phenol, guaiacol, guetol.

As regards the type of catalyst used, an at least difunctional acid can be used which corresponds to the following formula (II):



in which formula (II), R_1 represents a valency bond or an optionally substituted hydrocarbon radical containing 1 to 40 carbon atoms.

To be more exact, in formula (II), R_1 symbolises a substituted or non-substituted hydrocarbon radical which can be a linear or branched, saturated or unsaturated acyclic aliphatic radical; a monocyclic or polycyclic, saturated, unsaturated, or aromatic carbocyclic radical; a monocyclic or polycyclic, saturated, unsaturated or aromatic heterocyclic radical.

~~The carrier compounds with at least two carboxylic functions of general formula (II) in which R_1 represents a valency bond or a divalent radical preferably having 1 to 15 carbon atoms are quite particularly suitable for implementation of the process according to the invention.~~

~~The carrier compounds with at least two carboxylic functions of general formula (II) in which R_1 represents a linear or branched, saturated or unsaturated aliphatic residue are particularly well suited for use of the process according to the invention.~~

To be more exact, R_1 represents a linear or branched, acyclic aliphatic residue having preferably 1 to 12 carbon atoms, saturated or containing one or more unsaturations on the chain, generally 1 to 3 unsaturations which can be single or conjugated double bonds, or triple bonds.

The hydrocarbon chain can optionally be:

(1) - interrupted by one of the following groups called Y:

$$\begin{array}{c} -\text{O}- ; -\text{CO}- ; -\text{COO}- ; -\text{N}(\text{R}_2)- ; -\text{CO}-\text{N}(\text{R}_2)- ; -\text{S}- ; -\text{SO}_2- \end{array}$$

in which formula R_2 represents hydrogen or a linear or branched alkyl radical having 1 to 4 carbon atoms, preferably a methyl or ethyl radical, or a radical of $-(\text{CH}_2)_p-\text{COOH}$ type in which p is a number between 1 and 5,

(2)

- and/or bearing one of the following substituents:

- OH; - COOH ; - CHO ; - NO₂ ; - CN ; - NH₂ ; - SH ; - X ; CF₃

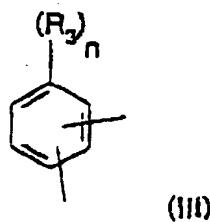
- NH - [(CH₂)_p - COOH] or - N -[(CH₂)_p - COOH]₂

with X representing a halogen atom, preferably a fluorine, chlorine or bromine atom, and p having the meaning given hereinabove.

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~~The carrier compounds with at least two carboxylic functions of general formula (II) in which R₁ represents a monocyclic or polycyclic hydrocarbon residue are also suitable for implementation of the process according to the invention.~~

R₁ preferably represents an aromatic hydrocarbon residue, and, in particular, a benzene residue corresponding to the general formula (III):



in which formula (III):

- n is an integer from 0 to 4, preferably from 0 to 3,
- R₃ represents one of the following groups or functions,
 - . a hydrogen atom,
 - . a linear or branched alkyl radical having from 1 to 4 carbon atoms,
 - . a linear or branched alkoxy radical having from 1 to 4 carbon atoms,
 - . a methylene or ethylene dioxy radical,
 - . a -CHO group,
 - . a phenyl or benzyl radical,
 - . a halogen atom,

Even more preferably, the compounds of formula (II) are selected in which the R₁ radical corresponds to formula (III) in which the R₃ radicals, which are identical or different, are a hydrogen atom, a methyl radical, a methoxy radical, a -CHO group.

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~~The carrier compounds having at least two carboxylic functions can correspond to general formula (II) in which the R₁ radical represents a polycyclic aromatic hydrocarbon divalent residue; the rings can form between themselves ortho-condensed, ortho- and peri-condensed~~

systems. More particularly, a naphthylenic residue can be mentioned; said rings being able to be substituted by 1 to 4 R_3 radicals, preferably by 1 to 3, R_3 having the meanings stated hereinabove for the substituents of the aromatic hydrocarbon residue of general formula (III).

Subt B10

In general formula (II) of the carrier compounds with at least two carboxylic functions, R_1 can also represent a carbocyclic residue which is saturated or which comprises 1 or 2 unsaturations in the ring, generally having from 3 to 7 carbon atoms, preferably 6 carbon atoms in the ring; said ring being able to be substituted by 1 to 5, preferably 1 to 3, R_3 radicals, R_3 having the meanings stated hereinabove for the substituents of the aromatic hydrocarbon residue of general formula (III).

As preferred examples of R_1 radicals the cyclohexane-diyl radicals can be mentioned which are optionally substituted by linear or branched alkyl radicals having 1 to 4 carbon atoms.

Subt B11

The carrier compounds with at least two carboxylic functions can also correspond to formula (II) in which R_1 represents a divalent radical constituted by a chain formation of two to four residues as defined hereinabove, an aliphatic residue, an aromatic residue, or a cycloaliphatic residue. These can be connected together by a valency bond or by a function group which can be, in particular, a group selected from the groups called Y.

Some examples of R_1 radicals are given hereinafter:

- CH₂-C₆H₄-;
- CH₂-CH₂-C₆H₄-;
- CH₂-O-C₆H₄-;
- CH₂-O-C₆H₄-;
- CH₂-O-C₆H₄-CH₂-;
- C₆H₄-C₆H₄-;
- C₆H₄-CH₂-C₆H₄-;
- C₆H₄-O-C₆H₄-;
- CH₂-C₆H₄-CH₂-C₆H₄-CH₂-

Subt B12

The following carrier compounds with at least two carboxylic functions can be mentioned, quite particularly, by way of catalysts which are suitable for the present invention:

- dicarboxylic aliphatic acids, such as:

- . oxalic acid
- . malonic acid
- . succinic acid
- . glutaric acid
- . adipic acid
- . 2,4-dimethyl adipic acid
- . pimelic acid
- . suberic acid
- . azelaic acid
- . sebatic acid
- . dodecane dioic acid
- . fumaric acid
- . maleic acid

- cycloalkanedicarboxylic acids, such as cyclohexane 1,4-dicarboxylic acid,

- aromatic dicarboxylic acids, such as:

- . phthalic acid
- . isophthalic acid
- . terephthalic acid
- . phenylenediacetic acid
- . naphthalene 1,5-dicarboxylic acid
- . naphthalene 1,6-dicarboxylic acid
- . 4,4'-diphenylcarboxylic acid
- . 3,3'-diphenylcarboxylic acid
- . bis(4-hydroxycarbonyl) phenyl oxide
- . bis(3-hydroxycarbonyl) phenyl oxide
- . 4,4'-dihydroxycarbonyl diphenylsulphone
- . 3,3'-dihydroxycarbonyl diphenylsulphone
- pyrimidine or imidazole dicarboxylic acids.

In the afore-mentioned list of dicarboxylic acids, the compounds preferably used are: oxalic acid, malonic acid, succinic acid, glutaric acid, adipic acid, sebatic acid, phthalic acid, isophthalic acid, terephthalic acid.

Aminopolycarboxylic acids are also perfectly well suited for implementation of the process according to the invention. As examples of aminopolycarboxylic acids suitable for use in the process of the invention mention can be made of the following, inter alia:

- ethylenediaminetetraacetic acid (E.D.T.A.)
- diethylenetriaminopentacetic acid (D.T.P.A.)
- nitrilotriacetic acid (N.T.A.)
- N-(2-hydroxyethyl)ethylene diaminetriacetic acid (H.E.D.T.A.)

Of the afore-mentioned aminopolycarboxylic acids, ethylenediaminetetraacetic acid is preferably selected.

According to the process of the invention, the reaction is carried out in the presence of an alkali metal hydroxide which can be sodium or potassium hydroxide.

For economic considerations, sodium hydroxide is preferably selected.

With regard to the concentrations and quantities of reagents to be used, the preferred conditions are defined hereinbelow.

In accordance with the process according to the invention a solution of glyoxylic acid is used. The concentration of said solution is not critical and can vary greatly, e.g. between 15 and 70% by weight. Preferably, commercial solutions are used, the concentration of which is approximately 50%.

According to the process of the invention, the glyoxylic acid is reacted on the excess of the hydroxylated aromatic compound of formula (I). The molar ratio between the hydroxylated aromatic compound of formula (I) and the glyoxylic acid varies between 1.5 and 4.0 and is preferably selected between 2.0 and 3.0.

The alkali metal hydroxide solution used has a concentration which is generally between 10 and 50% by weight. The concentration of the starting solution is not critical. However, as the concentration of the hydroxylated aromatic compound of formula (I) is advantageously low in the reaction medium, a dilute solution of alkali metal is used to carry out the dilution of the reaction medium.

The quantity of alkali metal hydroxide introduced into the reaction medium takes account of the quantity necessary to saponify the hydroxyl function of the hydroxylated aromatic compound of formula (I) and of the quantity necessary to saponify the carboxylic function of the glyoxylic acid.

If the hydroxylated aromatic compound of formula (I) has saponifiable functions other than the hydroxyl group, the quantity of alkali metal hydroxide necessary to saponify all the saponifiable functions which can be hydroxyl groups and/or COOH carboxylic functions is therefore introduced.

Generally, the quantity of alkali metal hydroxide can vary greatly and be equal to, or approximately equal to, the stoichiometry, or in excess. Generally, the quantity of alkali metal hydroxide varies between 80 and 120% of the stoichiometric quantity.

The concentration of the hydroxylated aromatic compound of formula (I) is preferably between 0.5 and 1.5 moles/litre, and, more particularly, in the region of 1 mole/litre.

With regard to the quantity of catalyst used, this is determined in such a way that the molar ratio between the catalyst and the hydroxylated aromatic compound of formula (I) is between 0.005 and 0.025, and preferably between 0.01 and 0.02.

The quantity of catalyst used, as expressed by the ratio between the number of moles of catalyst and the number of moles of glyoxylic acid is advantageously selected to be between 0.5 and 2.5%, preferably between 1 and 2%.

The preferred catalyst is oxalic acid.

The commercial solutions of glyoxalic acid can contain very low quantities of oxalic acid. The oxalic acid of the reaction can therefore be provided, in part, by the starting solution. In this case, it will be necessary to complete the quantity of oxalic acid by the addition of oxalic acid or of any other dicarboxylic acid in order that the afore-mentioned ratios are observed.

According to a preferred embodiment of the invention, a solution of glyoxylic acid is advantageously used containing between 0.6 and 3%, preferably between 1.2 and 2.6%, by weight of oxalic acid, expressed in relation to the weight of glyoxylic acid.

The reaction temperature is advantageously selected between 20°C and 60°C, and preferably between 30°C and 40°C.

The process according to the invention is carried out at atmospheric pressure, but under a controlled inert gas atmosphere, preferably nitrogen or rare gases, in particular nitrogen.

A preferred practical embodiment of the invention will be given hereinafter.

The solution of glyoxylic acid and catalyst, and, in parallel, the solution of alkali metal hydroxide employed in a quantity necessary to salify the COOH function are introduced into a reaction medium containing the hydroxylated aromatic compound of formula (I), water and the alkali metal hydroxide in a quantity necessary to salify the hydroxyl group and other possible salifiable functions of the compound of formula (I).

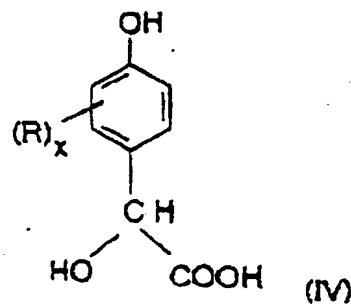
The reaction medium is kept under agitation and at the temperature selected within the afore-mentioned range for a variable period ranging from 1 to 10 hours.

Another variant of the execution of the invention consists in adding the reaction catalyst not to the aqueous solution of glyoxylic acid but simultaneously with the hydroxylated aromatic compound of formula (I).

At the end of the reaction, the optionally substituted p-hydroxymandelic acid obtained is separated in salified form using conventional separation techniques, in particular crystallisation.

The process according to the invention is quite particularly well suited when an aqueous solution of glyoxylic acid is used containing monofunctional acids, such as formic and glycolic acid, and, in particular, when acetic acid is present in a concentration which is variable between 0.1 and 3%.

The process according to the invention results in the production of optionally substituted p-hydroxymandelic compounds which can be represented by the following formula (IV):



in which formula (IV) R and x have the meaning given in formula (I).

These products are particularly worthwhile since they are intermediate products, which, *inter alia*, allow hydroxyarylacetic acids to be obtained by reduction and hydroxyarylglyoxylic (= hydroxyaryl α -oxo acetic) acids or hydroxyaromatic aldehydes to be obtained by oxidation.

A preferred use of the invention is the preparation of hydroxyaromatic aldehydes by oxidation of the compounds of formula (IV) obtained according to the invention.

The oxidation of the compounds of formula (IV) can be carried out according to the techniques described in writings. Thus, reference can be made to P. HEBERT [Bull. Soc. Chim. France, 27, p. 45-55 (1920)] and to NAGAI SHIGEKI et al [JP-A 76/128934]. The oxidation is generally carried out using oxygen or air under pressure, in the presence of an appropriate catalyst, such as derivatives of chromium, cobalt, copper, vanadium or osmium, for example.

Therefore, the invention allows easy access to 4-hydroxy benzaldehyde and to vanillin and its analogues, e.g. 3-ethyl, 3-isopropyl vanillin, by oxidation respectively of p-hydroxymandelic acid and of 3-methoxy p-hydroxymandelic acids, and 3-ethoxy p-hydroxy-mandelic or 3-isopropoxy p-hydroxymandelic acids.

The following examples illustrate the invention, without however limiting it.

In the examples, the percentages stated are expressed by weight.

The abbreviations mentioned in the examples have the following meanings:

Conversion (TT) =
$$\frac{\text{number of moles of guaiacol converted}}{\text{number of moles of guaiacol introduced}}$$

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Yield (RR) =
$$\frac{\text{number of moles of mandelic acid formed}}{\text{number of moles of mandelic acid introduced}}$$

X

Selectivity (RT) =
$$\frac{\text{number of moles of mandelic acid formed}}{\text{number of moles of guaiacol converted}}$$

Example 1

- 600 g distilled water,
- 91.6 g (0.687 mol) of a 30% aqueous soda solution,
- 93 g (0.750 mol) guaiacol

are introduced into a 1-litre glass reaction vessel equipped with a double jacket, a pH electrode, a temperature probe, a condenser, an inert gas supply, and a mechanical agitation device.

An inert atmosphere is established, and the reaction mixture is brought to 35°C, and 50.7 g (0.380 mol) of a 30% by weight aqueous soda solution and 55.2 g of a 50% by weight aqueous glyoxylic acid solution are added simultaneously over a period of 2 hours. Oxalic acid is added with the glyoxylic acid introduced in such a quantity that it represents 0.75% by weight of the glyoxylic acid solution.

The glyoxylic acid solution introduced contains 0.3% oxalic acid, 0.9% lower carboxylic acids such as acetic acid, and formic acid and glycolic acid in a respective quantity of less than 0.1%.

The reaction mixture is maintained at 35°C for 2 hours.

At the end of the reaction, the reaction products are analysed using high performance liquid chromatography.

The results obtained were as follows:

- conversion:
 - . TT = 47.3%
- 4-hydroxy 3-methoxy mandelic acid:
 - . RR = 79.7%
 - . RT = 84.2%
- 2-hydroxy 3-methoxy mandelic acid:
 - . RR = 4.8%
 - . RT = 5.1%
- 2-hydroxy 3-methoxy 1,5-dimandelic acid:
 - . RR = 8.0%
 - . RT = 4.0%

Comparative Example 2

Example 1 is reproduced, except that no oxalic acid is introduced.

The results obtained were as follows:

- conversion:
 - . TT = 46.1%
- 4-hydroxy 3-methoxy mandelic acid:
 - . RR = 76.9%
 - . RT = 83.0%
- 2-hydroxy 3-methoxy mandelic acid:
 - . RR = 5.1%
 - . RT = 5.5%
- 2-hydroxy 3-methoxy 1,5-dimandelic acid:
 - . RR = 7.5%
 - . RT = 4.1%

Example 3

In this example, Example 1 is reproduced except that a 50% solution of glyoxylic acid is used containing 0.4% by weight of oxalic acid.

The results obtained are as follows:

- conversion:
 - . TT = 48%
- 4-hydroxy 3-methoxy mandelic acid:
 - . RR = 79.3%
 - . RT = 83.1%
- 2-hydroxy 3-methoxy mandelic acid:
 - . RR = 5.6%
 - . RT = 5.8%
- 2-hydroxy 3-methoxy 1,5-dimandelic acid:
 - . RR = 8.0%
 - . RT = 4.2%

Examples 4 to 8

In the following set of examples Example 1 is reproduced except that different types of dicarboxylic acids are used, such as malonic acid, succinic acid and E.D.T.A.

The glyoxylic acid solution used contains oxalic acid at a rate of 0.09%, lower carboxylic acids such as acetic acid at a rate of 1%, formic acid and glycolic acid at respective quantities of less than 0.3%.

All the conditions of the examples and the results obtained are indicated in Table I.

Table 1

Ref. ex.	Dicarboxylic acid (%)	TT	RR			RT		
			para	ortho	di	para	ortho	di
4	-	45.2	77.5	4.9	7.6	84.3	5.3	4.1
5	oxalic acid (2.0%)	47.8	80.1	4.8	8.1	83.0	5.0	4.2
6	malonic acid (2.0%)	46.1	80.2	5.2	7.6	84.8	5.5	4.1
7	succinic acid (1.9%)	48.0	81.4	5.6	8.0	85.0	5.8	4.2
8	E.D.T.A. (1.5%)	44.5	80.6	4.9	7.7	88.0	5.4	4.2

* = dicarboxylic acid expressed in molar percent in relation to the glyoxylic acid.

Examples 9 to 11

In the following examples, the quantity of oxalic acid used in the glyoxylic acid solution is increased.

The procedure of Example 1 is followed, and a 50% glyoxylic acid solution is used, the composition of which is given in Examples 4 to 8.

The results obtained are indicated in the following table:

Table II

Ref. ex.	Oxalic acid *	TT	RR			RT		
			para	ortho	di	para	ortho	di
9	1.00	45.2	79.5	5.1	7.6	86.7	5.6	4.1
10	1.9	47.8	80.1	4.8	8.1	83.0	5.0	4.2
11	1.78	46.1	76.5	4.8	8.0	83.0	5.2	4.3

* = concentration of oxalic acid in % by weight in the glyoxylic acid solution.

In said table, the abbreviations, "ortho", "para" and "di" mean :

- 4-hydroxy 3-methoxy mandelic acid = para

- 2-hydroxy 3-methoxy mandelic acid = ortho
- 2-hydroxy 3-methoxy 1,5 dimandelic acid = di